airpointer® NOx module





Chemiluminescence

n = 3 n = 2 n = 1 $\Delta E = hv$ NO_2*

It is a two-part reaction:

1) Ozone and NO reaction. A dioxygen molecule and an "excited" NO₂ molecule is formed

$$O_3 + NO \rightarrow NO_2^* + O_2$$

2) This "excited" NO_2 molecule returns to its ground state, releasing its excess energy as a photon (h_V) whose center wavelength is about 1100 nm.

$$NO_2^* \rightarrow NO_2 + hv @1100 nm$$





Chemiluminescence: quenching

Another way for the excited NO_2 to get rid of its energy is by colliding with another molecule (X).

$$NO_2^* + X \rightarrow NO_2 + X^*$$

To minimize the chance of collision, the reaction chamber is kept under vacuum ($P \approx 0.3$ atm).





How is NO measured?

$$NO_2^* \rightarrow NO_2 + hv$$

•Intensity of emitted light is proportional to NO concentration and measured by a PMT

Only NO can be measured directly



How is NO_2 measured?

- A molybdenum catalyst, heated to ~325 °C, is used to convert NO₂ to NO
- Reduction of NO₂ to NO:

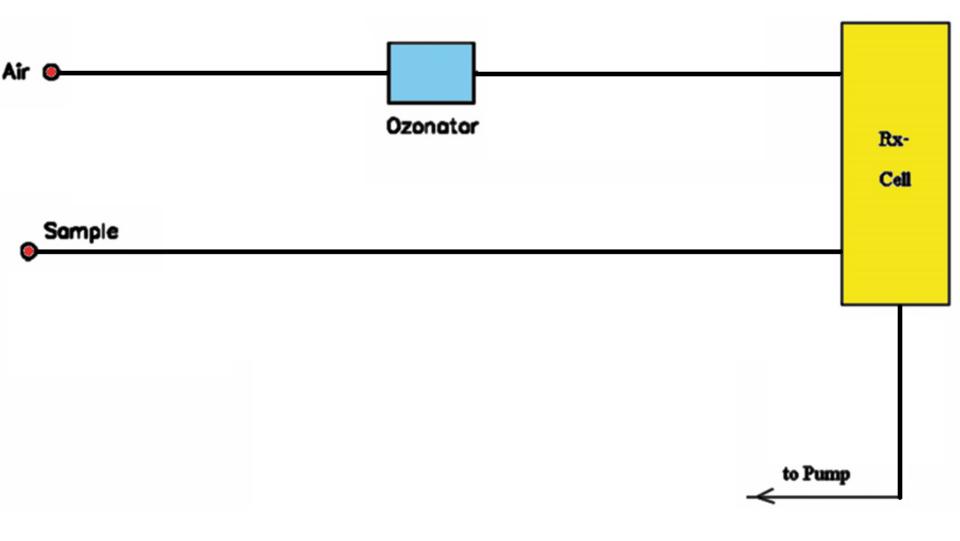
$$3 \text{ NO}_2 + \text{Mo} \rightarrow 3 \text{NO} + \text{MoO}_3$$

- Thus in the NOx-cycle, the sum of NO + NO₂ is measured
- Calculation for NO_2 : $c(NOx) c(NO) = c(NO_2)$



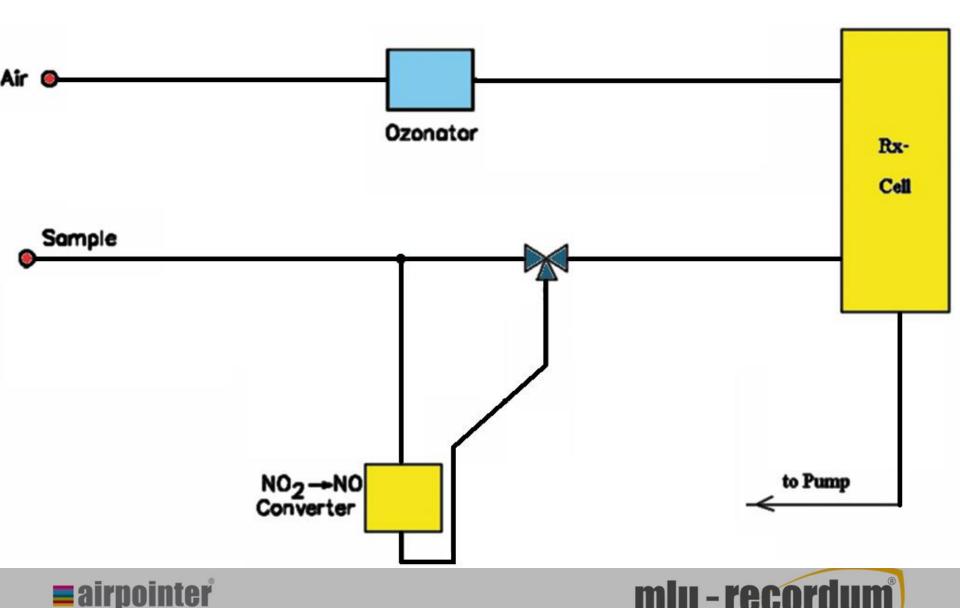


Flow schematic (ND)









Problems and solutions

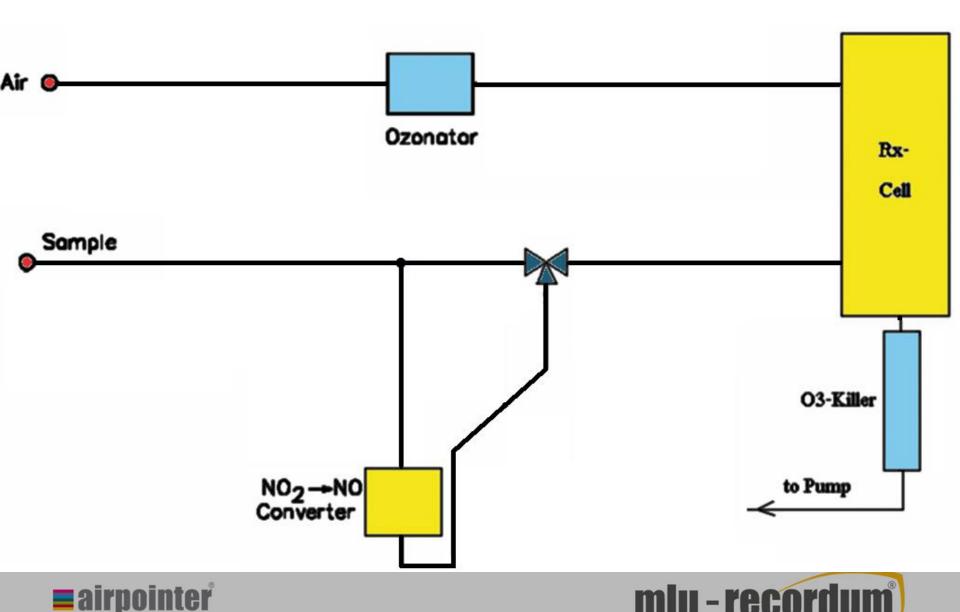
- O₃ is corrosive and toxic
 → Catalytic O₃ killer for low ozone exhaust
- NOx and NO cycle last 8" each: [NO₂] = [NOx] [NO]
 →Delay loop to minimize NO₂-artefacts

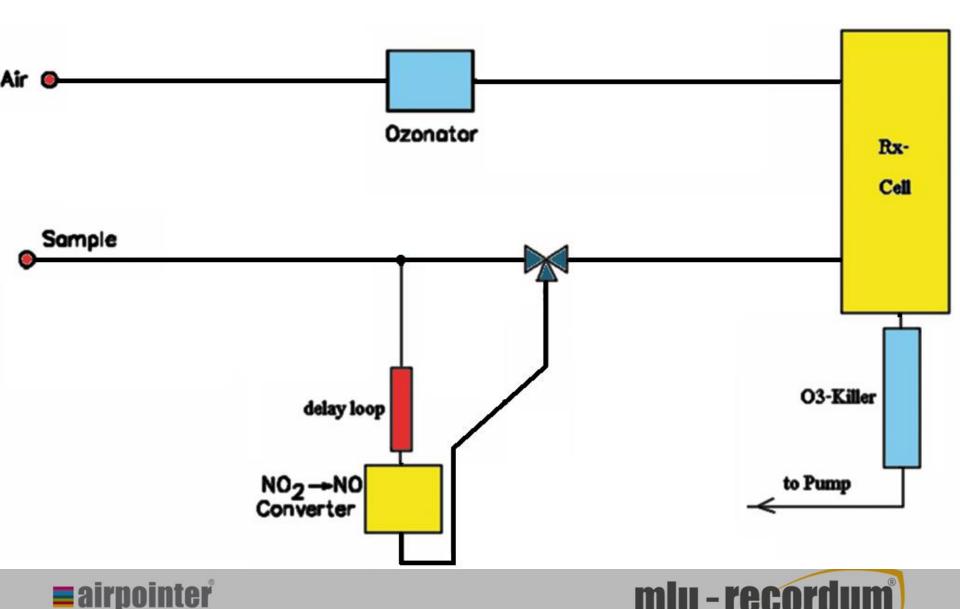
Ozonator generates unstable compounds (radicals)
 →O₃ cleanser to reduce interfering radicals

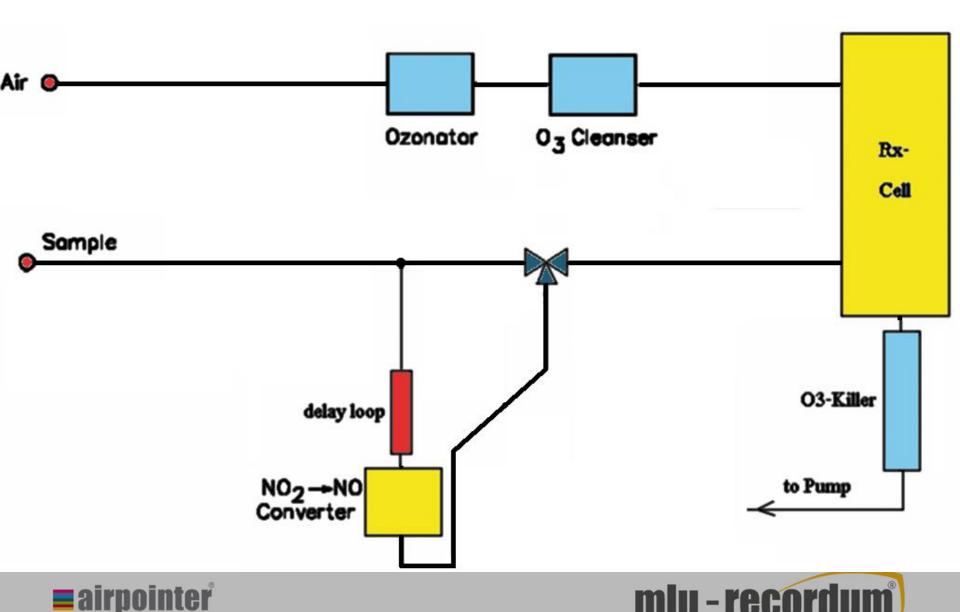
Humidity interference
 → Sample dryer

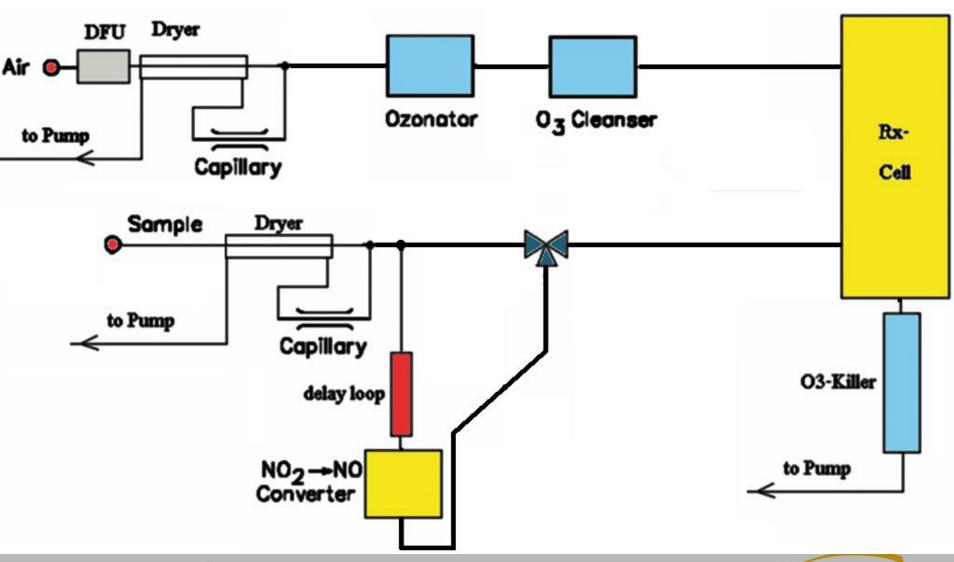






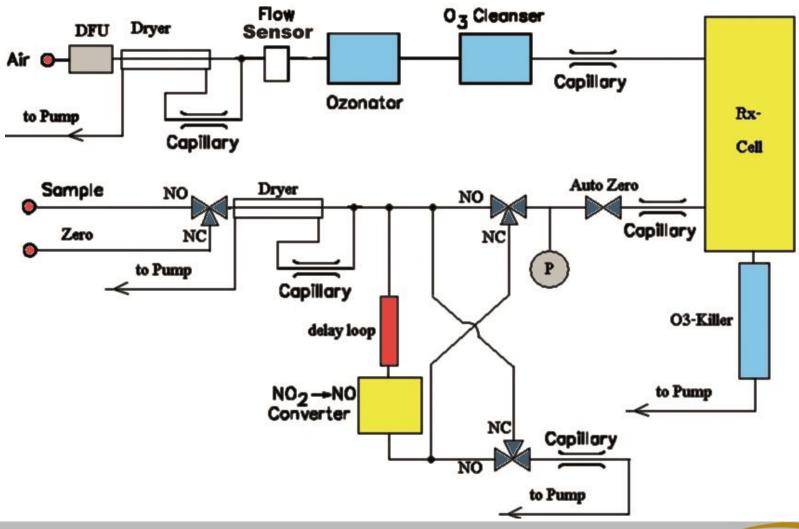






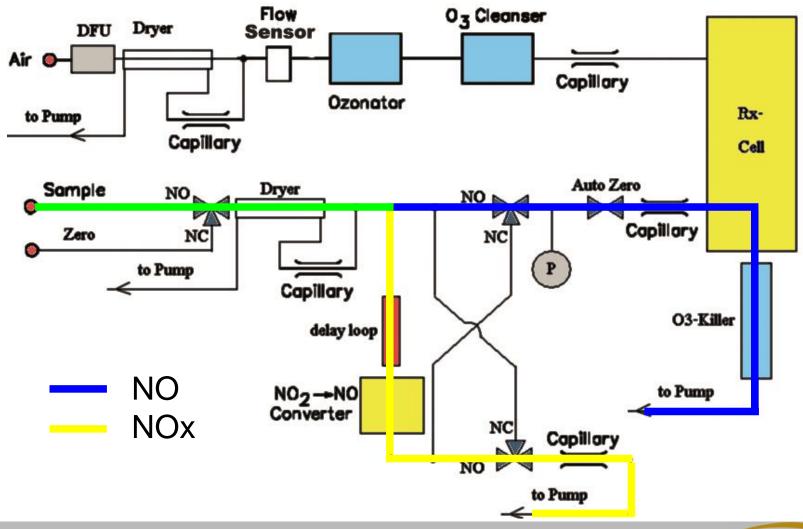






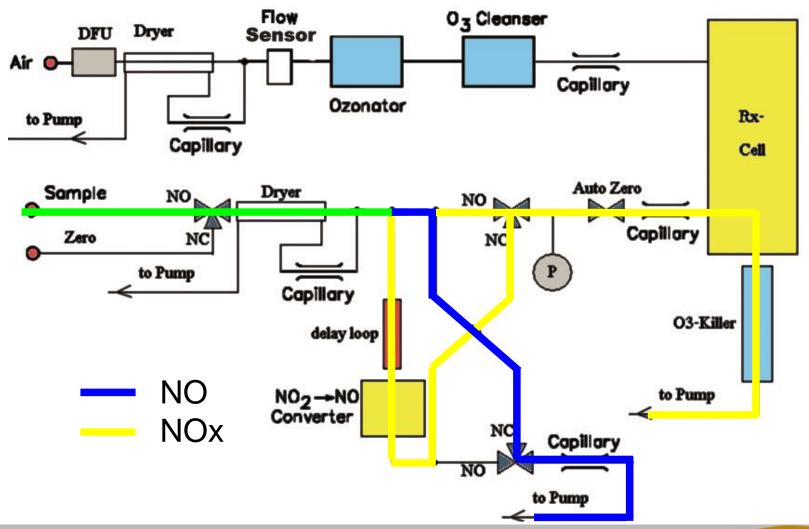








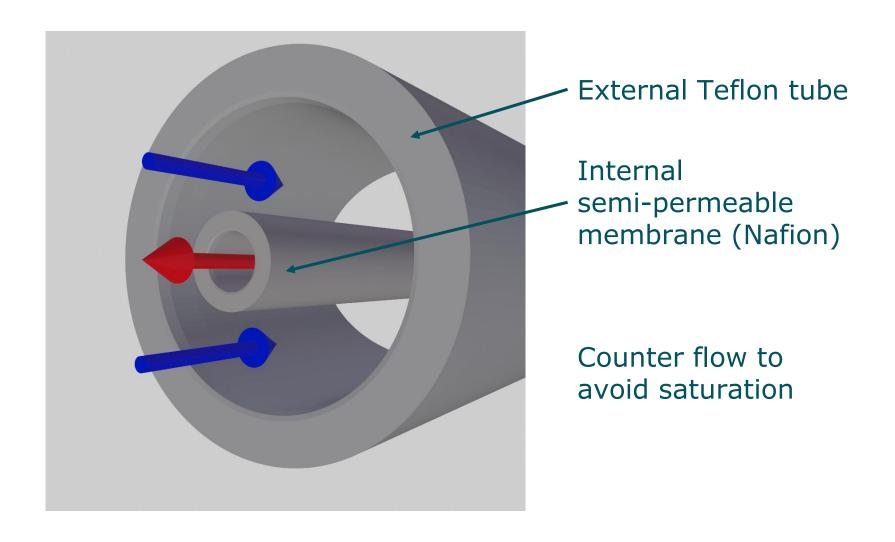








Permapure dryer







Reaction chamber

- Gold plated for maximized sensitivity (improved reflectivity).
- Chamber is under vacuum to prevent quenching so better stability and increased sensitivity.
- Chamber is heated to stabilize flow and thereby improve reproducibility

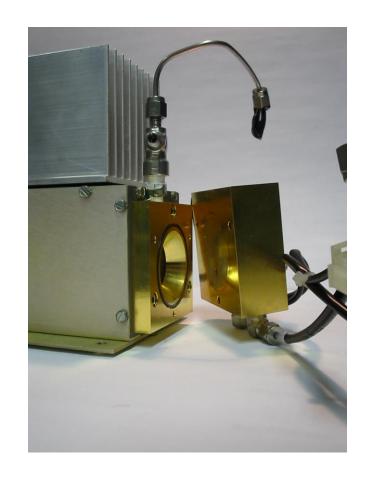






Photo multiplier tube

 Cooled to -2°C to increase sensitivity and stability







 $NH_3 \rightarrow NO_2$ converter before the $NO_2 \rightarrow NO$ converter





Parameters

Unit

ppb

Actual NOx Values O3Generator: ON

Parameter

NO

no calibration active next automatic calibration cycle starts: 20160909 02:00:00

Value

0.3

NO2	1.4		ppb			000						
NOx 1.7			ppb				000					
NO_all 0.3	ppb	NO_raw	(0.3	ppb	NOStdDev	0.23		NO_Avg (300 s	sec)	0.4	ppb
NO2_all 1.4	ppb	NO2_raw		1.4	ppb	NO2StdDev	0.30		NO2_Avg (300	sec)	1.0	ppb
NOx_all 1.7	ppb	NOx_raw	1.7		ppb	NOxStdDev	0.25	NOx_Avg (30		sec)	1.3	ppb
PMTSigNO	158.9		Hz		PMTSigNOx		384.3			Hz		
PMTSigAuto0 159.2			Hz									
PressNO 902.3		mbar		RCellPressNO		330.2			mbar			
PressNOx 902.7			mbar		RCellPressNOx		331.2			mbar		
FlowNOx 1145.6		ml/min		FlowO3Gen		96.6			ml/min			
Fan_NOx	3390		rpm		HVPS	HVPS_NOx		-636		V		
PMTTemp -2.0		°C		PowerToPeltier		78.8			%			
MolyT 325.4			°C		PowerToMoly		25.8			%		
RCelIT 50.2			°C		PowerToRCell		39.1			%		
PermT	49.8		°C		PowerToPerm		25.6		%			
NO Time Constant nr values to TC:				1		StdDev last 10 samples:				0.16		
NO2 Time Constant nr values to TC:				1		StdDev last 10 samples:				0.28		
NOx Time Constant nr values to TC:				1		StdDev last 10 samples:				0.32		
NO Slope:				1.027		NO Offset:				-0.237		
NOx Slope:				1.030		NOx Offset:				-0.186		
NO2 CE:				1.000		Gain 100 /Offset 5.0 /HVset 630.0 V				O3Gen ON		

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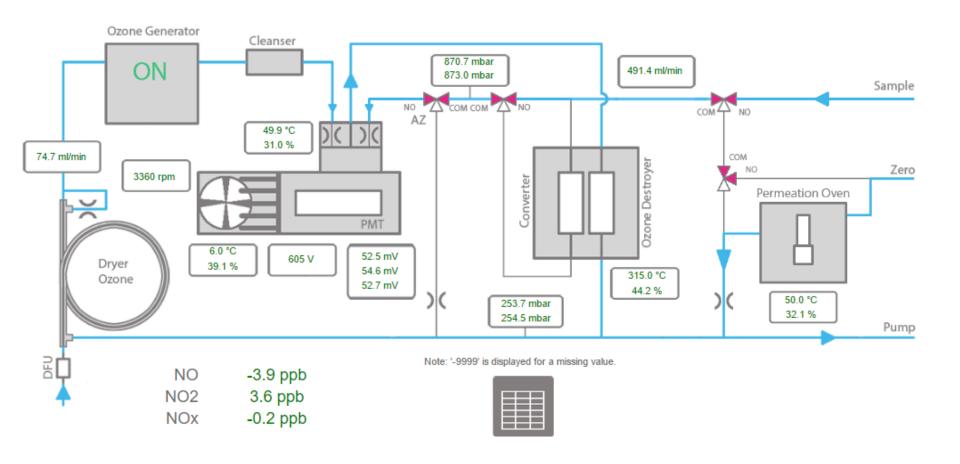




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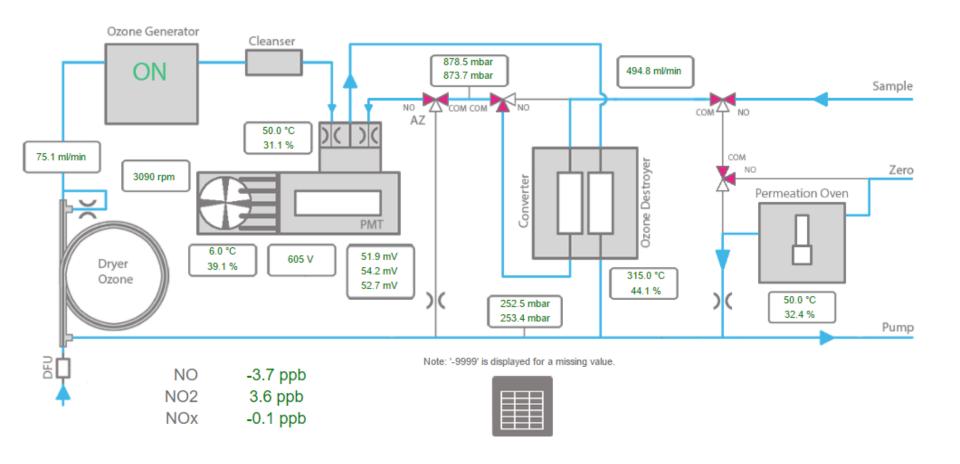
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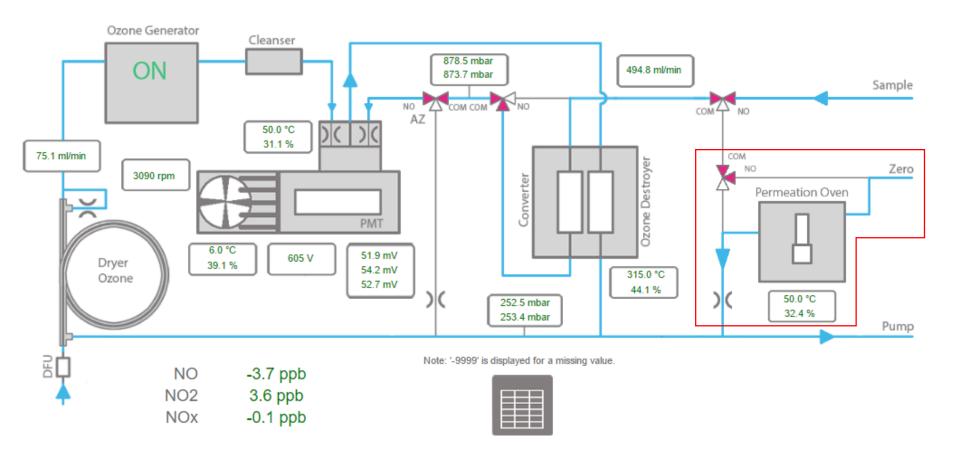
Parameters







Span check







Calibration

For NO:

- Using external zero air and external NO cylinder

For NO₂:

- Using external zero air and external Gas Phase Titration (GPT) module to generate NO_2 from NO and O_3





Preventive maintenance

- Change DFU filters (once a year)
- Test the molybdenum converter efficiency (during each calibration)
- Change o-rings (once a year)
- Clean reaction cell (once a year)

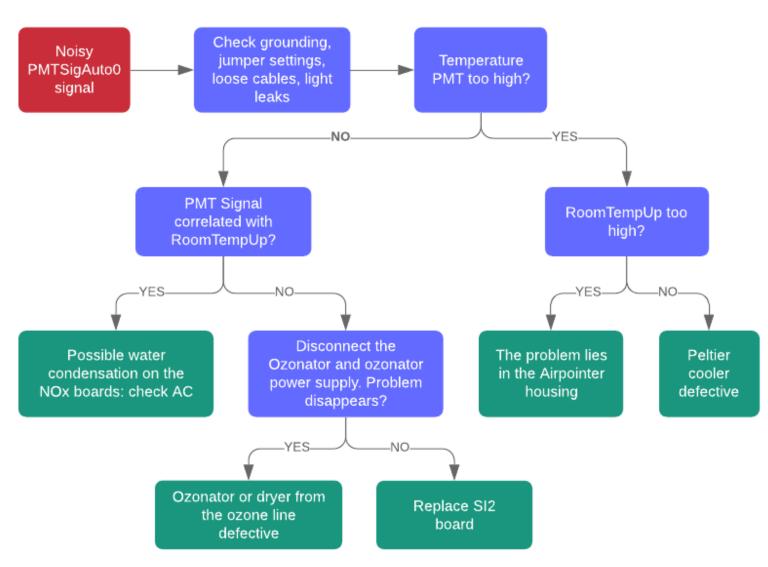
Full schedule available here:

https://www.airpointer.tech/maintenance-schedule/





Troubleshooting NOx







Thank you for your attention!



